

## Aqua(1,10-phenanthroline)(thiodiglycolato)cobalt(II)

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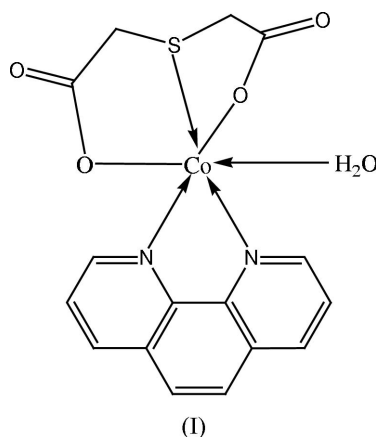
## Key indicators

Single-crystal X-ray study  
T = 295 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
R factor = 0.038  
wR factor = 0.091  
Data-to-parameter ratio = 15.9For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title cobalt compound,  $[\text{Co}(\text{C}_4\text{H}_4\text{O}_4\text{S})(\text{C}_{12}\text{H}_8\text{N}_2)(\text{H}_2\text{O})]$ , the distorted octahedron around the Co atom is formed by the O atom from a water molecule, two N atoms from the heterocycle, two carboxyl O atoms and an S atom from the dianion. Hydrogen bonds extend the structure into a two-dimensional network.

## Comment

The S atom of thiodiglycolic acid may be coordinated to metal atoms. However, only a few metal–thiodiglycolate complexes have been reported (Bonomo *et al.*, 1982; Baggio *et al.*, 1996, 1999; Kopel *et al.*, 2003; Grirrane *et al.*, 2003).

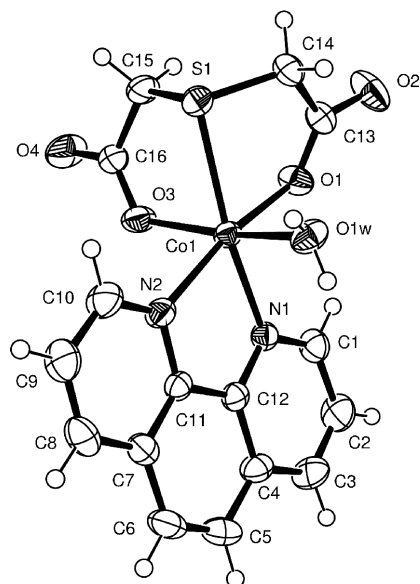


In the title compound, (I), the coordination polyhedron around the Co atom can be described as a distorted octahedron consisting of two N-atom donors from a 1,10-phenanthroline, one O atom from the water molecule and three donors from the thiodiglycolate ligand (Fig. 1 and Table 1). The flexible dicarboxylate dianion is converted to a rigid ligand when the S-atom donor coordinates to the  $\text{Co}^{\text{II}}$  atom, giving rise to the formation of two five-membered chelate rings. Both rings display a twist conformation. Each carboxyl group is coordinated in monodentate fashion to the cobalt centre. Both uncoordinated carboxyl O atoms form hydrogen bonds with water molecules, resulting in a two-dimensional hydrogen-bonding network (Fig. 2 and Table 2).

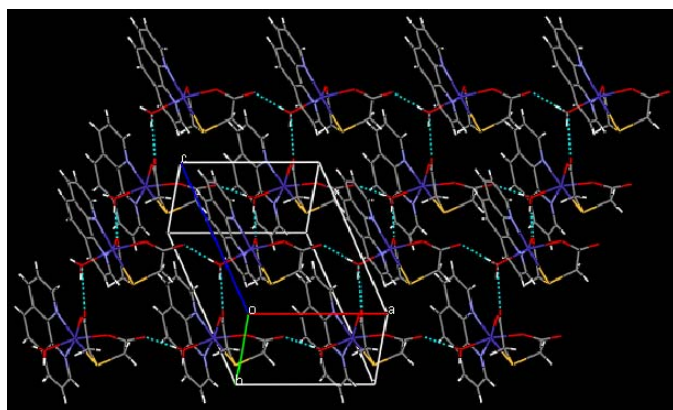
## Experimental

A mixture of cobalt(II) acetate tetrahydrate (0.0747 g, 0.30 mmol), thiodiglycolic acid (0.0452 g, 0.30 mmol), 1,10-phenanthroline (0.0595 g, 0.30 mmol) and water (10 ml) was heated at 393 K for 24 h in a 20 ml Teflon-lined stainless steel autoclave. After cooling, blue block-shaped crystals of (I) were obtained.

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**Figure 1**  
ORTEP-3 view (Farrugia, 1997) of the title compound. Displacement ellipsoids are drawn at the 50% probability level.



**Figure 2**  
View of the two-dimensional hydrogen-bonding network of (I), with hydrogen bonds shown as dashed lines.

#### Crystal data

[Co(C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>S)(C<sub>12</sub>H<sub>8</sub>N<sub>2</sub>)(H<sub>2</sub>O)]  
*M<sub>r</sub>* = 405.28  
 Monoclinic, *P*<sub>2</sub><sub>1</sub>/*n*  
*a* = 8.0127 (9) Å  
*b* = 22.524 (3) Å  
*c* = 9.733 (1) Å  
 $\beta$  = 113.420 (1)°  
*V* = 1611.9 (3) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.670 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 3581 reflections  
 $\theta$  = 2.8–24.7°  
 $\mu$  = 1.23 mm<sup>-1</sup>  
*T* = 295 (2) K  
 Block, blue  
 0.14 × 0.14 × 0.09 mm

#### Data collection

Bruker SMART APEX area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 2002)  
*T<sub>min</sub>* = 0.784, *T<sub>max</sub>* = 0.898  
 18 142 measured reflections

3688 independent reflections  
 2997 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.050  
 $\theta_{\max}$  = 27.5°  
*h* = -10 → 10  
*k* = -29 → 29  
*l* = -12 → 12

#### Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.038  
*wR* (*F*<sup>2</sup>) = 0.091  
*S* = 1.02  
 3688 reflections  
 232 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0432P)^2 + 0.4148P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.38 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.27 \text{ e } \text{Å}^{-3}$

**Table 1**

Selected geometric parameters (Å, °).

Co1—O1	2.033 (2)	Co1—N1	2.114 (2)
Co1—O3	2.038 (2)	Co1—N2	2.149 (2)
Co1—O1W	2.084 (2)	Co1—S1	2.5191 (7)
O1—Co1—O3	94.78 (8)	O1W—Co1—N2	88.34 (8)
O1—Co1—O1W	92.07 (8)	N1—Co1—N2	77.91 (7)
O3—Co1—O1W	166.26 (8)	O1—Co1—S1	80.51 (5)
O1—Co1—N1	91.64 (7)	O3—Co1—S1	81.51 (5)
O3—Co1—N1	100.60 (7)	O1W—Co1—S1	87.90 (5)
O1W—Co1—N1	91.08 (7)	N1—Co1—S1	172.03 (5)
O1—Co1—N2	169.55 (7)	N2—Co1—S1	109.95 (5)
O3—Co1—N2	87.07 (7)		

**Table 2**

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1W—H1W1...O2 <sup>i</sup>	0.84 (1)	1.93 (1)	2.764 (3)	170 (3)
O1W—H1W2...O4 <sup>ii</sup>	0.85 (3)	1.82 (3)	2.662 (3)	179 (3)

Symmetry codes: (i)  $x - \frac{1}{2}, \frac{3}{2} - y, z - \frac{1}{2}$ ; (ii)  $x - 1, y, z$ .

The H atoms bonded to C atoms were positioned geometrically and included in the refinement using the riding-model approximation [*C*—H = 0.93 Å for CH and *C*—H = 0.97 Å for CH<sub>2</sub>, and *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(C)]. The water H atoms were located in a difference Fourier map and refined with distance restraints of O—H = 0.85 (1) Å and with fixed isotropic displacement parameters of *U*<sub>iso</sub>(H) = 0.05 Å<sup>2</sup>.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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